#### Transweb Educational Services Pvt Ltd

-147, Sector -6 Noida, UP - 201301, <u>www.askiitians.com</u> Email: info@askiitians.com Tel: 1800-2000-838

### askIITians

ENGINEERING | MEDICAL | FOUNDATION

#### Class: 12

Subject: chemistry Topic: Halo alkanes No. of Questions: 20 Duration: 60 Min Maximum Marks: 60

- 1. Molecular formula of a dihalide is  $C_2H_4Br_2$ . This dihalide when treated with aqueous KOH, acetaldehyde is formed. Dihalide may be
  - a. Ethylene dibroamide
  - b. Ethylidene bromide
  - c. Acetylene bromide
  - d. None of these

#### Sol: B

A dihalogen derivative , ethylidene dibromide  $(CH_3CHBr_2)$  on hydrolysis gives a dihydric alcohol. Since the two - OH groups are found on the same carbon atom, the molecule is unstable. Hence it eliminates a mole)cule of water giving an aldehyde.  $CH_3CHBr_2 \xrightarrow{KOH} CH_3CH(OH)_2 \xrightarrow{-H_2O} CH_3CHO$ 

- 2. Isopropy1 chloride can be obtained by the action of
  - a. HC1 on acetone
  - b. Chlorine on acetaldehyde
  - c. PC1₅ on 1- propanol
  - d. PC1<sub>5</sub> on 2-propanol

Sol: D

- 3. n-Propy1 bromide on treating with ethanolic potassium hydroxide produces
  - a. propane
  - b. propene
  - c. propyne
  - d. propanol

Sol: B

```
\mathrm{CH}_3 \text{ - } \mathrm{CH}_2 \text{ - } \mathrm{CH}_2 \text{ - } \mathrm{Cl} + \mathrm{KOH} \xrightarrow{\text{Alcoholic}} \mathrm{CH}_3 \text{ - } \mathrm{CH} = \mathrm{CH}_2 + \mathrm{KCl} + \mathrm{H}_2\mathrm{O}
```

Transweb Educational Services Pvt Ltd B -147, Sector -6 Noida, UP - 201301, <u>www.askiitians.com</u> Email: info@aski<u>itians.com</u>

Tel: 1800-2000-838

### askiltians Engineering | medical | foundation

- 4. Anisole can be prepared by the action of methy1 iodide on sodium phenate. The reaction is known as
  - a. Reamer-Tiemann's reaction
  - b. Williamson's reaction
  - c. Etard's reaction
  - d. Schotten-Baumann reaction

Sol: B fact

- 5. Viny1 chloride can be prepared by the action of alcoholic KOH on
  - a. Ethylene dicholoride
  - b. Ethyl chloride
  - c. Allyl chloride
  - d. Benzy1 chloride

Sol: A fact

- 6. Ethyl iodide reacts with AgNO<sub>2</sub> to give
  - a. Nitroethane
  - b. Ethane
  - c. Ethyl nitrite
  - d. Ethylene

Sol: A

- 7. Which of the following compounds will produce an isocyanide on treatment with alkyl halide?
  - a. NaCN
  - b. KCN
  - c. AgCN
  - d. All

Sol: C

NaCN and KCN give cyanides while AgCN gives the isocyanide

 $R \textbf{-} X + AgCN \longrightarrow R \textbf{-} N \equiv C + AgX$ 

-  $N \equiv C$  group is called isocyanide

- 8. Gem dihalides on hydrolysis gives
  - a. Only aldehydes
  - b. Only ketones
  - c. Aldehydes or ketones

## askiltians

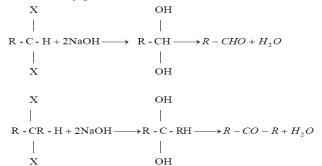
Transweb Educational Services Pvt Ltd

147, Sector -6 Noida, UP - 201301, <u>www.askiitians.com</u> Email: info@askiitians.com Tel: 1800-2000-838

d. Acids

Sol: C

Gem dihalides are dihalo alkanes where both the halogen atoms are attached to the same carbon atom. If halogen is attached to terminal carbon atom. On hydrolysis they give aldehydes. If halogens are not attached to the terminal carbon atoms, they give ketones



- 9. The alkyl halide having highest boiling points is
  - a. CH₃F
  - b. CH₃Br
  - c. CH₃I
  - d. CH₃Cl

Sol: C

For the given alkyl group higher the atomic mass of halogen more is the boiling point

- 10. Which among the following alkyl halide is hydrolysed by  $S_N$  l mechanism
  - a. Methyl bromide
  - b. Propyl iodide
  - c. Neo- Pentyl chloride
  - d. Tertiary butyl bromide

Sol: D

The order of reactivity for  $S_N$  mechanism is  $3^\circ>2^\circ>1^\circ$ . Among a compound given except the 4 th one remaining all are  $1^\circ$  alkyl halide. Hence option 4 is correct

- 11. Which among the following isomeric alkyl halides have least boiling point
  - a. CH<sub>3</sub>- CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-Br
  - b. CH<sub>3</sub>-CH-CH<sub>2</sub>-CH<sub>2</sub>-Br | CH<sub>3</sub>
  - c.  $CH_3$ -CH CH – $CH_3$



Tel: 1800-2000-838

$$| | CH_3 Br CH_3 Br CH_3 CH_3 CH_3 - C-CH_3 d. Br d.$$

Sol: D Higher the branching, lesser is the boiling point

- 12. Darzen's process is not applicable for the preparation of alkyl bromide because
  - a. Darzen's process involves PCL<sub>5</sub>
  - b. SOBr<sub>2</sub> is unstable
  - c. HBr is unstable
  - d. None of these

Sol: B fact

- 13. Catalyst used during antimarkonwnikoff s addition of HBr to alkene is
  - a. Ni
  - b. Benzoyl peroxide
  - c. Pt
  - d. AICI<sub>3</sub>

Sol: B fact

- 14. Which of the following does not react with benzene in the presence of anhydrous AlCl<sub>3</sub>?
  - a. C<sub>6</sub>H₅Cl
  - b. C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>Cl
  - c. CH<sub>3</sub>COCI
  - d.  $C_2H_5Br$

Sol: A

Chlorobenzene is very inactive. Hence it does not answer the Friedel-Crafts reaction mentioned in the question. CH<sub>3</sub> COCI reacts with benzene to form acetophenone. This reaction is called Friedel-Crafts acylation

- 15. Ethyl bromide reacts with KCN to form
  - a. Ethyl cyanide
  - b. Ethyl isocyanide
  - c. Ethyl isocyanate
  - d. Ethyl cyanate

Sol: A

 $C_2H_5Cl + KCN \longrightarrow C_2H_5 - C \equiv N + KCl$  $-C \equiv N$  group is called cyanide group

# askiltians

-147, Sector -6 Noida, UP - 201301, <u>www.askiitians.com</u> Email: info@askiitians.com Tel: 1800-2000-838

- 16. 1, 2-dichloroethane when boiled with alcoholic potash gives
  - a. Acetic acid
  - b. Formic acid
  - c. Potassium acetate
  - d. Vinyl chloride

Sol: D

Alcoholic potash is a dehydrohalogenating agent. In the first stage vinyl chloride is formed. In the next stage ethyne is formed

- 17. Which statement is not true regarding benzyl chloride
  - a. it is an aromatic compound with substitution in the side chain
  - b. it gives a white precipitate with alcoholic AgNO<sub>3</sub>
  - c. it is less reactive than vinyl chloride
  - d. it undergoes nucleophilic substitution reactions

Sol: C fact

- 18. The best reagent for converting ethanol to chloroethane is
  - a. PCl<sub>5</sub>
  - b. PCl<sub>3</sub>
  - c. SOCl<sub>2</sub>
  - d. HCl in the presence of ZnCl<sub>2</sub>

Sol: C

```
C_2H_5OH + SOCl_2 \longrightarrow Cl_2H_5Cl + SO_2 \uparrow + HCl \uparrow
```

By products, SO2 and HCl are gases and can be easily separated. In other cases at least one of the products is a liquid. Hence purificati on is difficult

- 19. Isopropyl chloride is heated with sodium in dry ether. The product is
  - a. 2, 3-dimethylbutane
  - b. Pentane
  - c. Hexane
  - d. methyl pentane

Sol: A

$$\begin{array}{ccc} CH_3 & CH_3 & CH_3 \\ | & | & | \\ 2CH_3 - CH - Cl + 2Na \longrightarrow CH_3 - CH - CH - CH_3 + 2NaCl \end{array}$$



Transweb Educational Services Pvt Ltd

147, Sector -6 Noida, UP - 201301, <u>www.askiitians.com</u> Email: info@askiitians.com Tel: 1800-2000-838

- 20. Monohalogen derivatives may be prepared by treating alcohols with
  - a. thionyl chloride
  - b. phosphorus halides
  - c. hydrohaloacids in the presence of anhydrous ZnCl<sub>2</sub>
  - d. any one of the above reagents

Sol: D

 $\begin{array}{l} \text{R-OH} + \text{SOCl}_2 \rightarrow \text{RCl} + \text{SO}_2 + \text{HCl} \\ \text{R-OH} + \text{PCl}_5 \rightarrow \text{RCl} + \text{POCl}_3 + \text{HCl} \\ \text{R-OH} + \text{HX} \rightarrow \text{RX} + \text{H}_2\text{O} \end{array}$