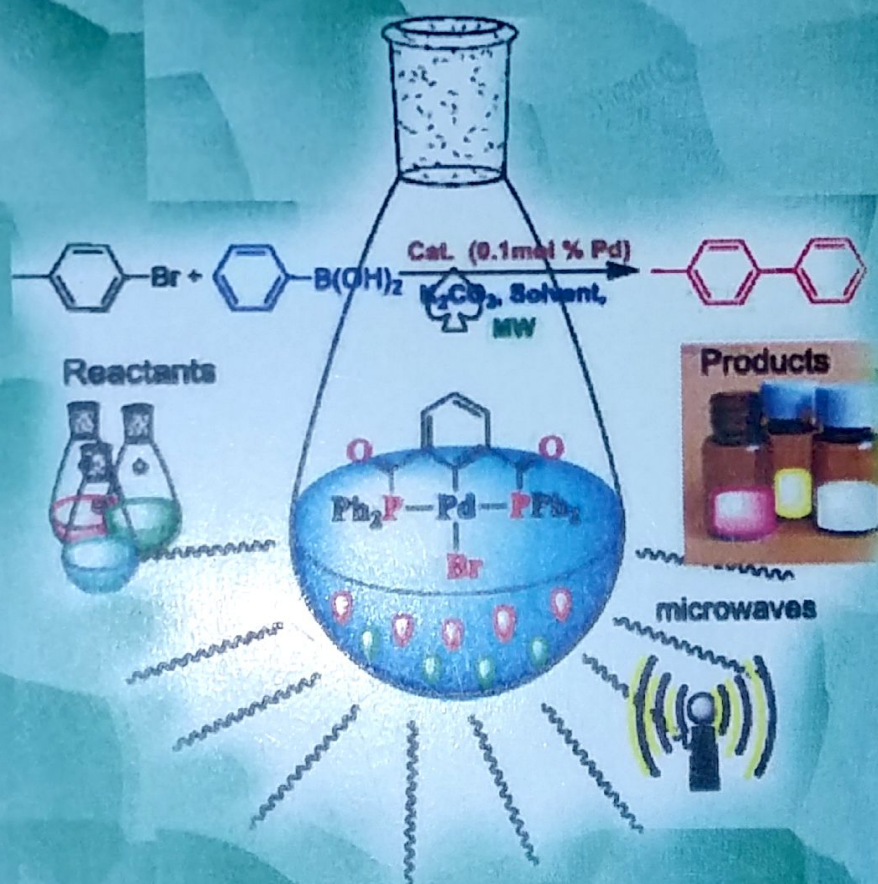


Organic Chemistry

Conversions



A. Shivakumar. M.Sc (Peradeniya)

CHEMISTRY

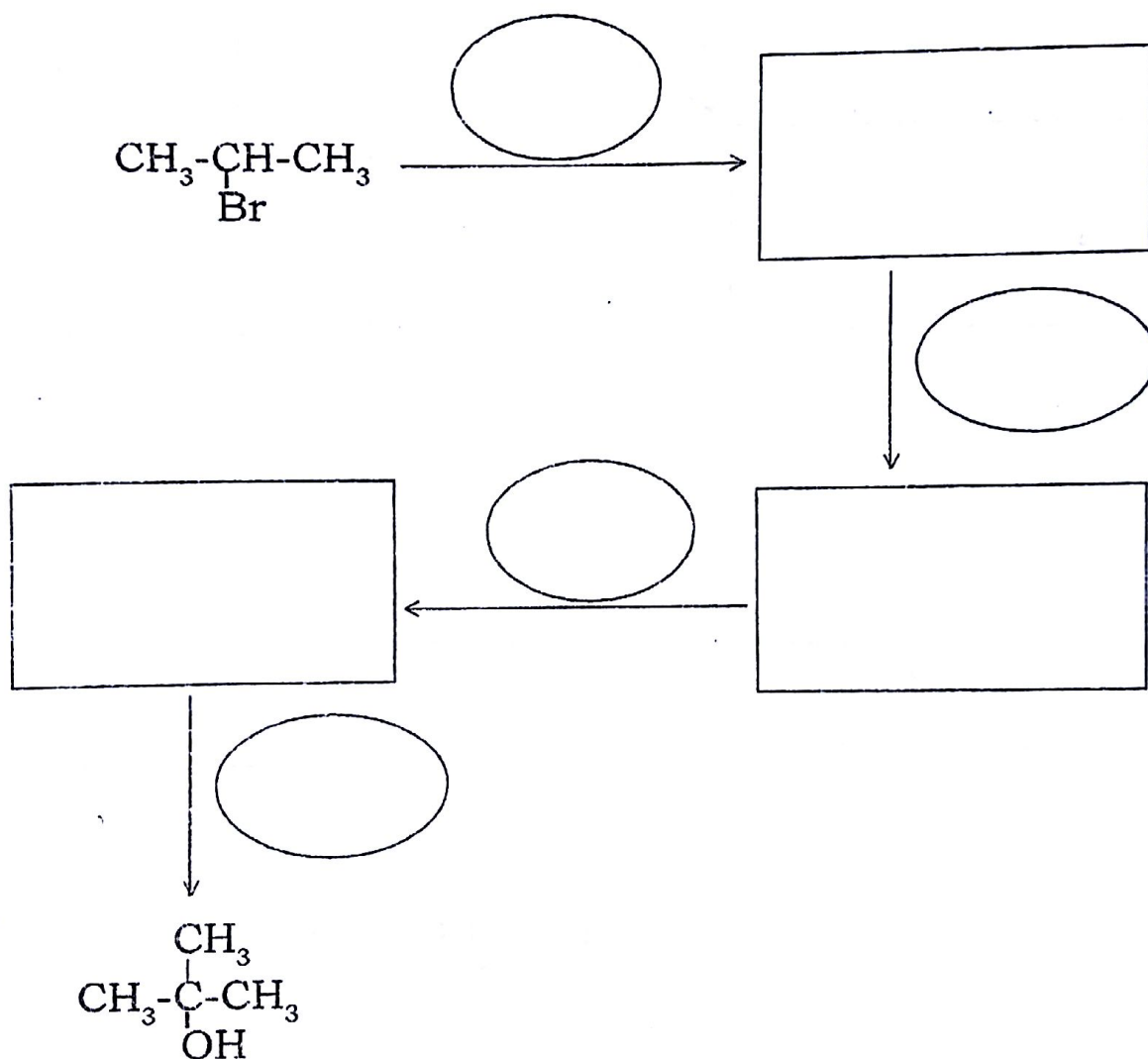
Organic Conversions

01. Selecting appropriate reagents and solvents **only from the list below**, show how you would synthesis the following compounds.

List of the reagent and solvent:

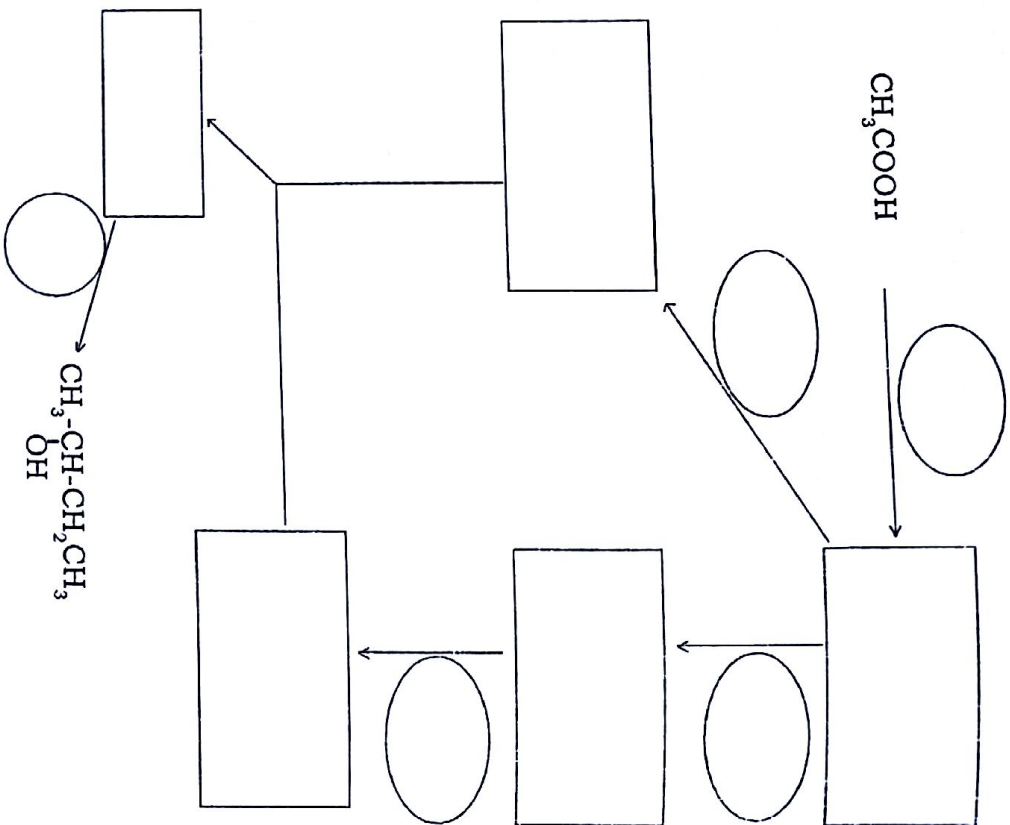
$K_2Cr_2O_7$, dil. H_2SO_4 , aq NaOH,

CH_3MgBr , water, dry $(C_2H_5OC_2H_5)$



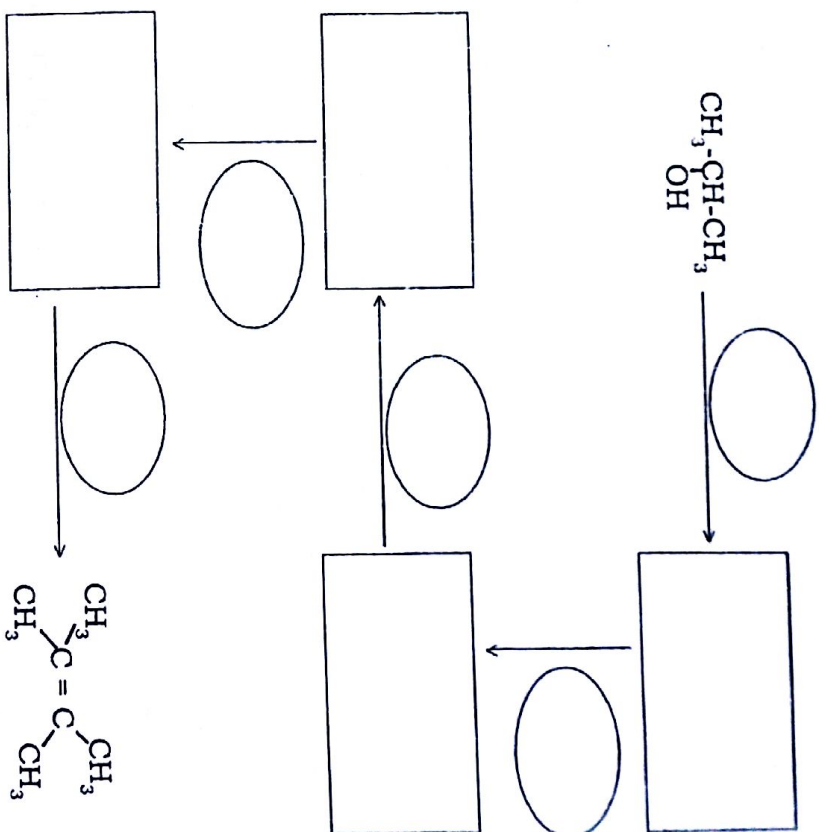
02. List of the reagent and solvent:

Mg, Br₂, PCl₅,
 LiAlH₄, aq NaOH, Pyridiniumchlorochromate(PCC)
 C₂H₅OH, water, dry (C₂H₅OC₂H₅)

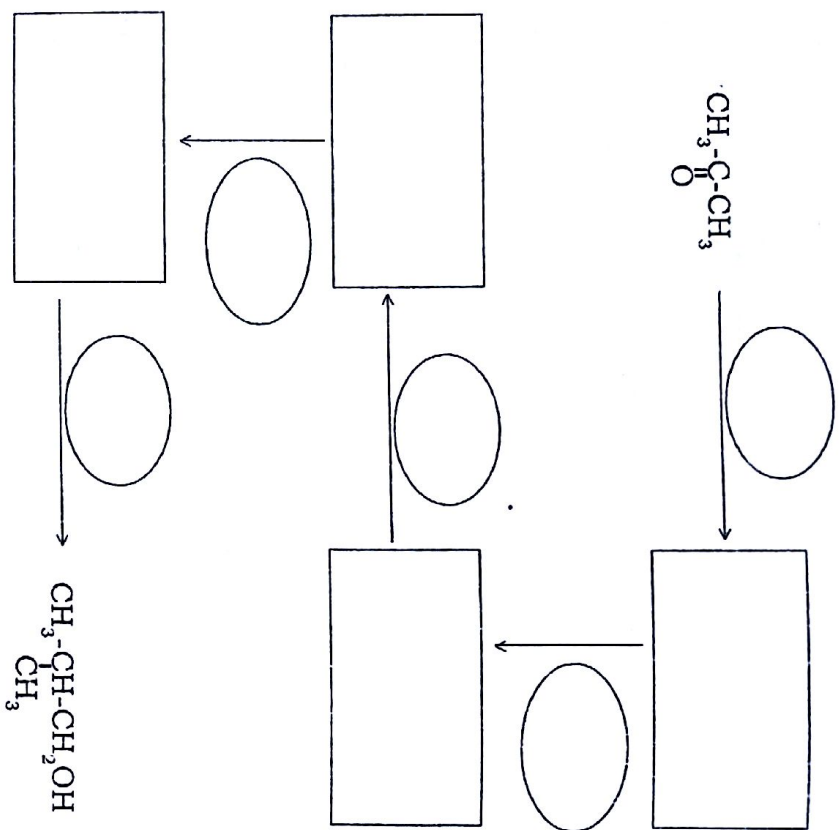


03. List of the reagent and solvent:

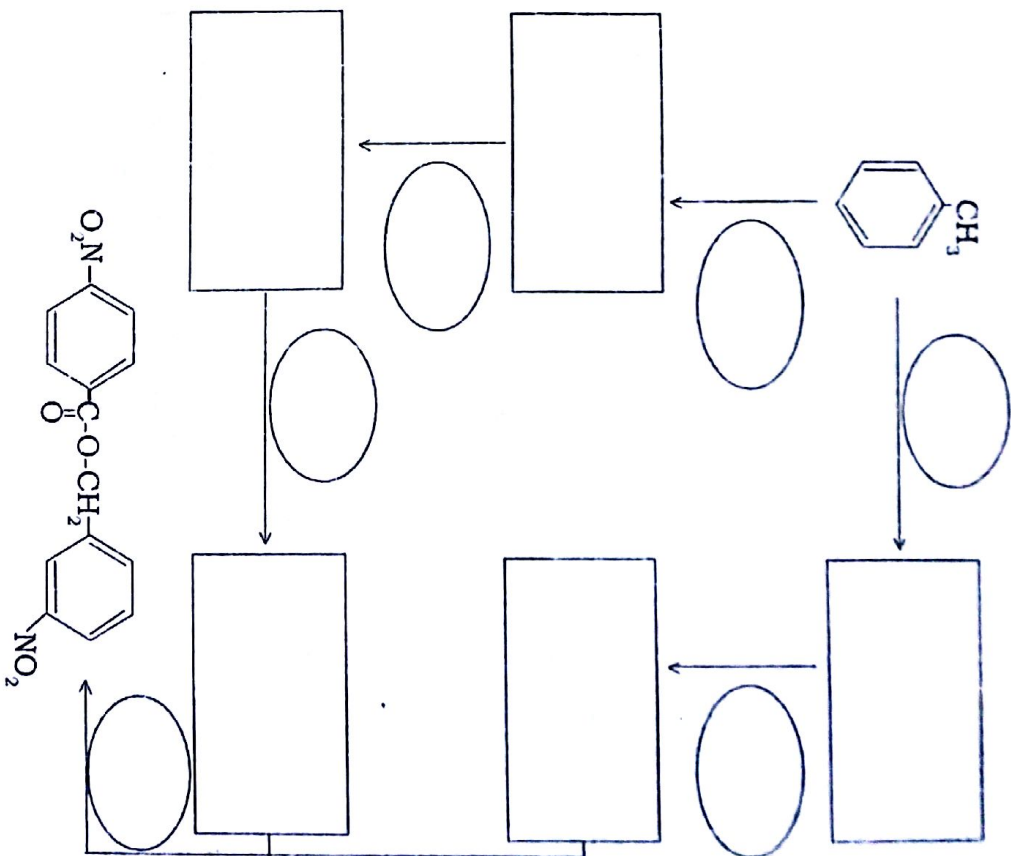
Mg, Br₂, PCl₅, HgSO₄, KOH
 K₂Cr₂O₇, Conc. H₂SO₄, aq NaOH,
 CH₃COCH₃, C₂H₅OH, water, dry (C₂H₅OC₂H₅)



04. List of the reagent and solvent:
 Mg, PCl₅, LiAlH₄, KCN, dil HCl
 C₂H₅OH, water, dry (C₂H₅OC₂H₅)

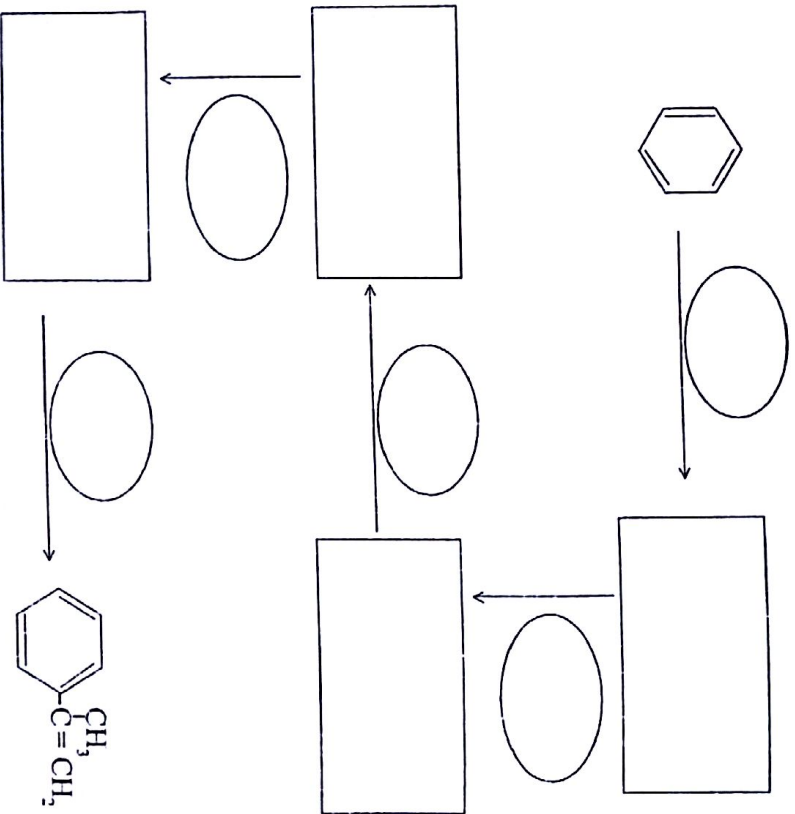


05 List of the reagent and solvent:
 PCl₅, AlCl₃, KCN, K₂Cr₂O₇, Conc. HNO₃, Conc. H₂SO₄,
 LiAlH₄, C₂H₅OH, water, dry (C₂H₅OC₂H₅)



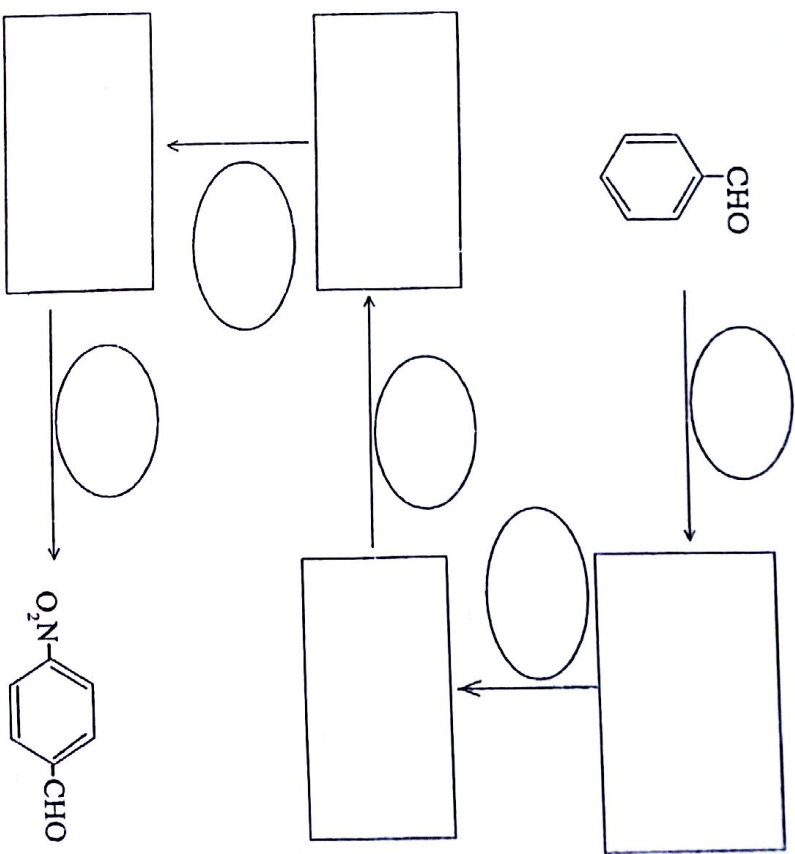
06. List of the reagent and solvent:

Mg, Fe, Br₂, PCl₅, AlCl₃
Conc. H₂SO₄, dil HCl, aq NaOH,
CH₃COCH₃, C₂H₅OH, water, dry (C₂H₅OC₂H₅)



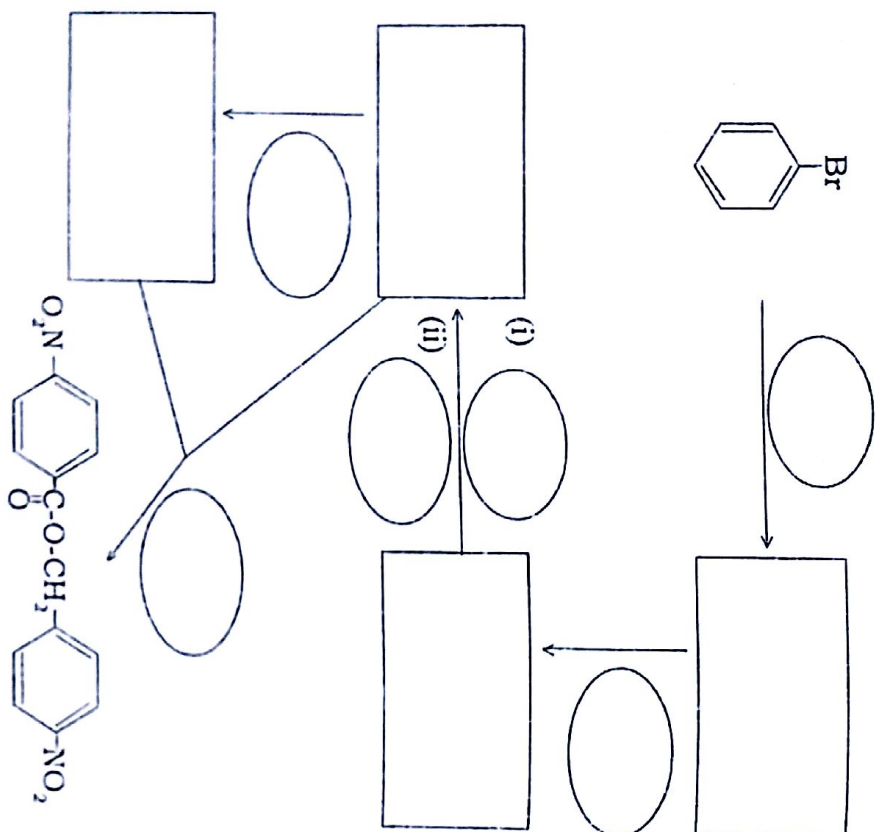
07. List of the reagent and solvent:

Zn, Fe, Hg, LiAlH₄, K₂Cr₂O₇, Conc. H₂SO₄, Conc. HCl
Conc. HNO₃, Pyridiniumchlorochromate(PCC),
C₂H₅OH, water, dry (C₂H₅OC₂H₅)



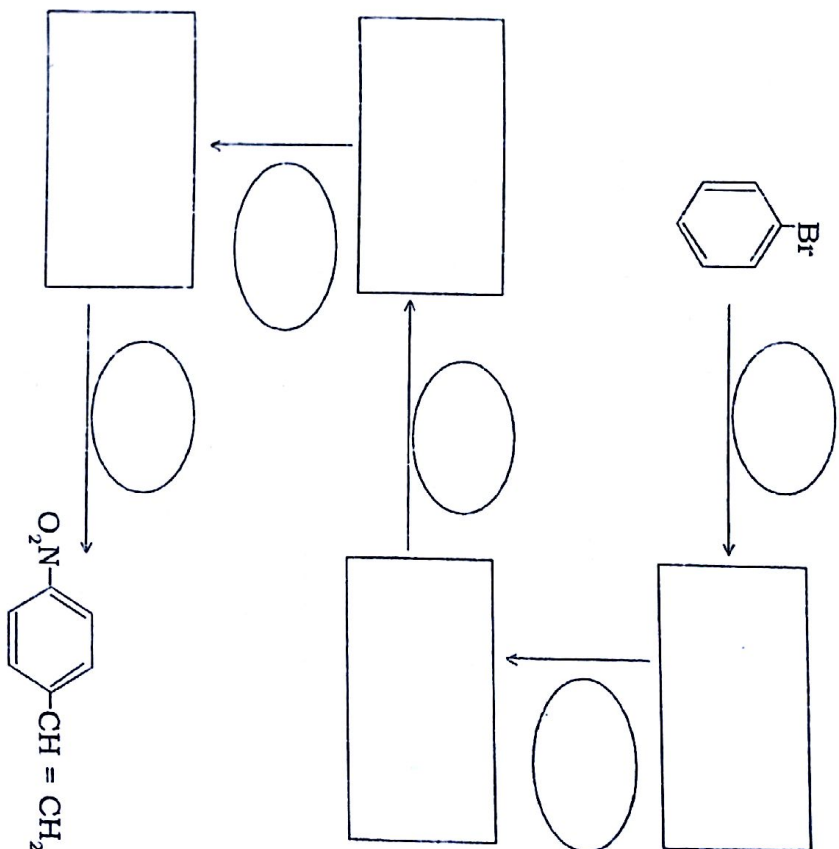
08. List of the reagent and solvent:

Mg, $K_2Cr_2O_7$, Conc. HNO_3 , Conc. H_2SO_4 , dil HCl
 CH_3CHO , C_2H_5OH , water, dry ($C_2H_5OC_2H_5$)

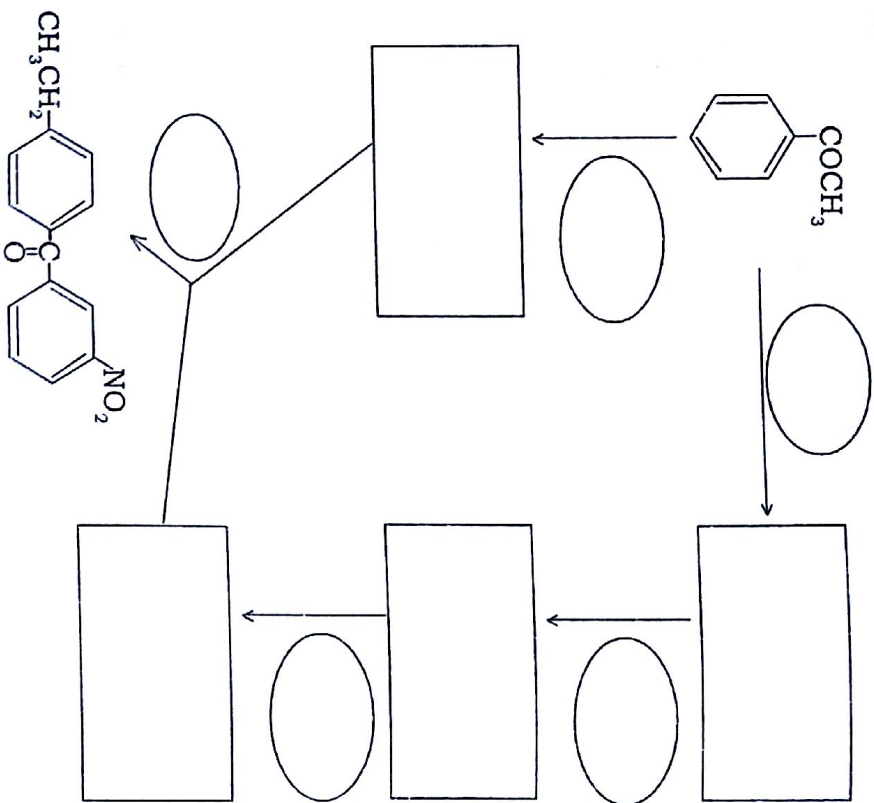


09. List of the reagent and solvent:

Mg, PCl_5 , Conc. HNO_3 , Conc. H_2SO_4 , dil HCl
 CH_3CHO , C_2H_5OH , water, dry ($C_2H_5OC_2H_5$)



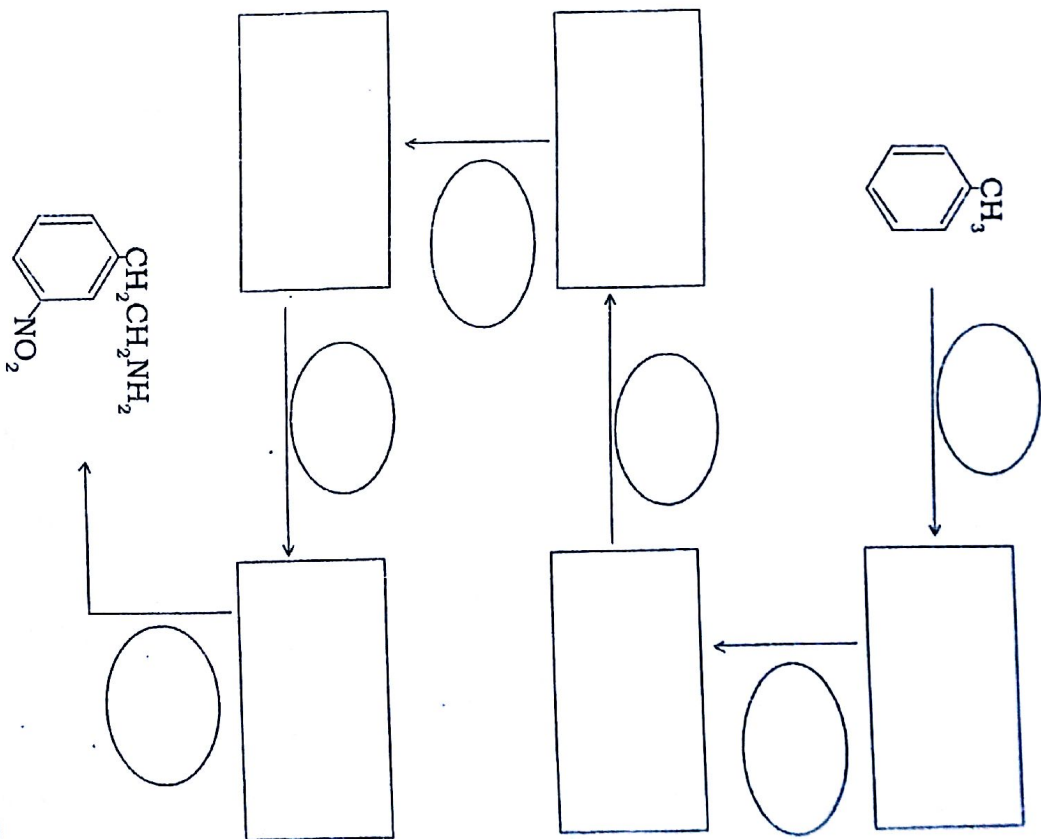
10. List of the reagent and solvent:
 Zn, Hg, PCl₅, K₂Cr₂O₇, Conc. HNO₃, Conc. H₂SO₄,
 AlCl₃, Conc. HCl, C₂H₅OH, water, dry (C₂H₅OC₂H₅)



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11. Mg, Zn, Pt, Fe, Hg
 Br₂, PCl₅, AlCl₃
 KCN, K₂Cr₂O₇, Conc. HNO₃, Conc. H₂SO₄, Conc. HCl
 aq NaOH,
 C₂H₅OH, water, dry (C₂H₅OC₂H₅)

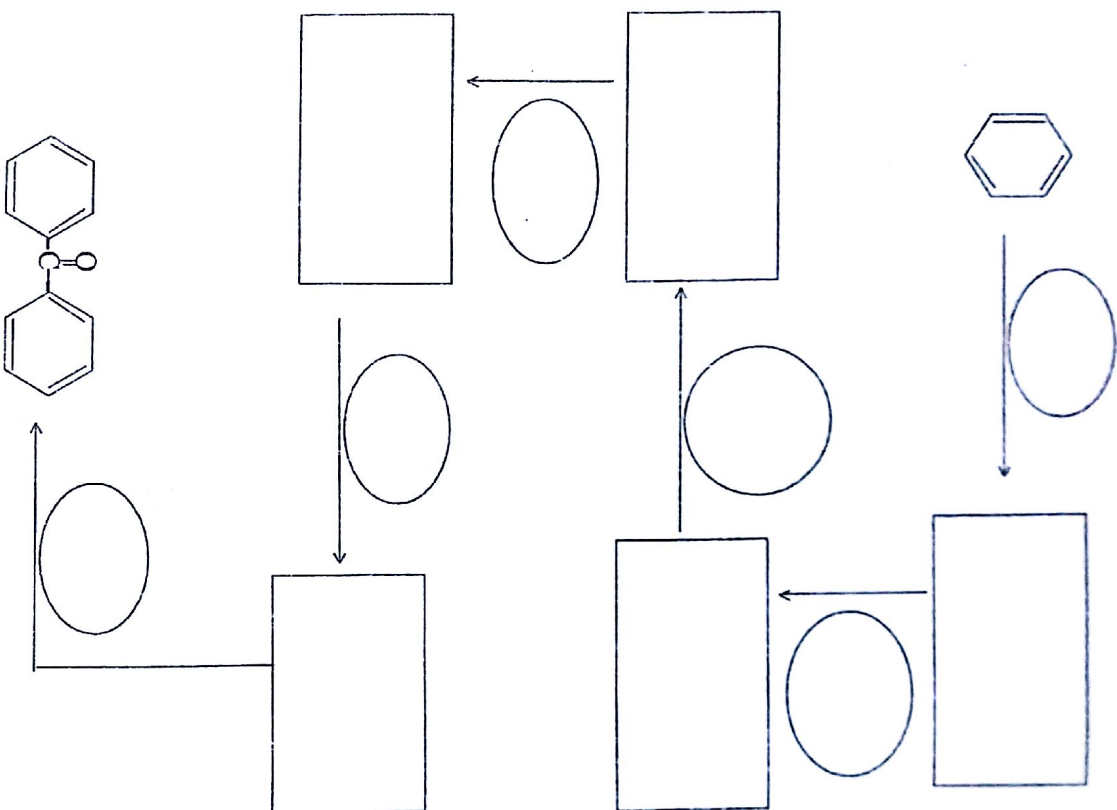


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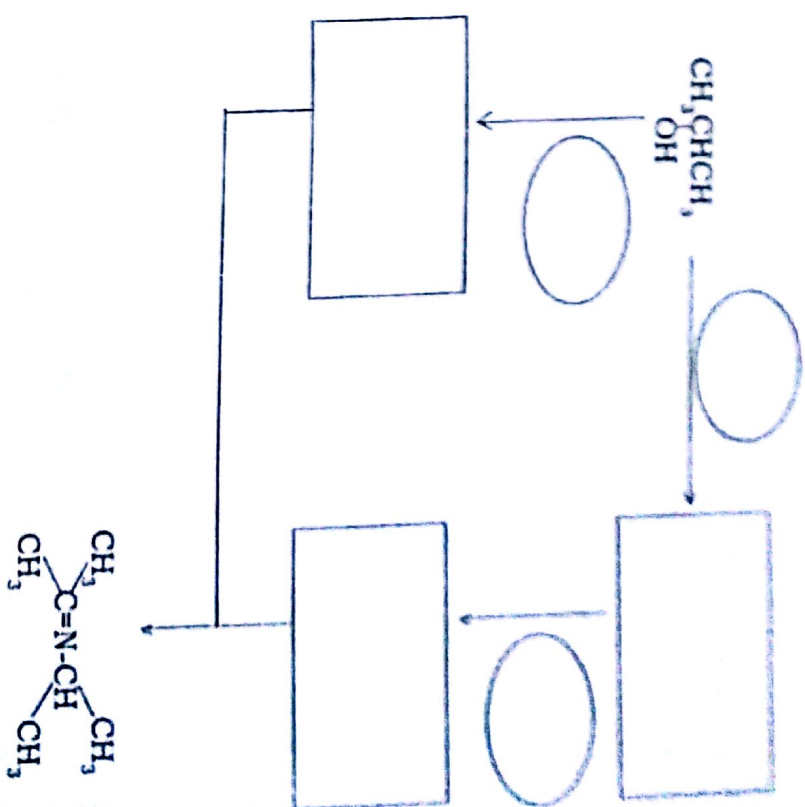
12. List of the reagent and solvent:

Mg, Fe, PCl₅, AlCl₃,
 Br₂, K₂Cr₂O₇, dil H₂SO₄,
 HCHO, C₂H₅OH, water, dry (C₂H₅OC₂H₅)

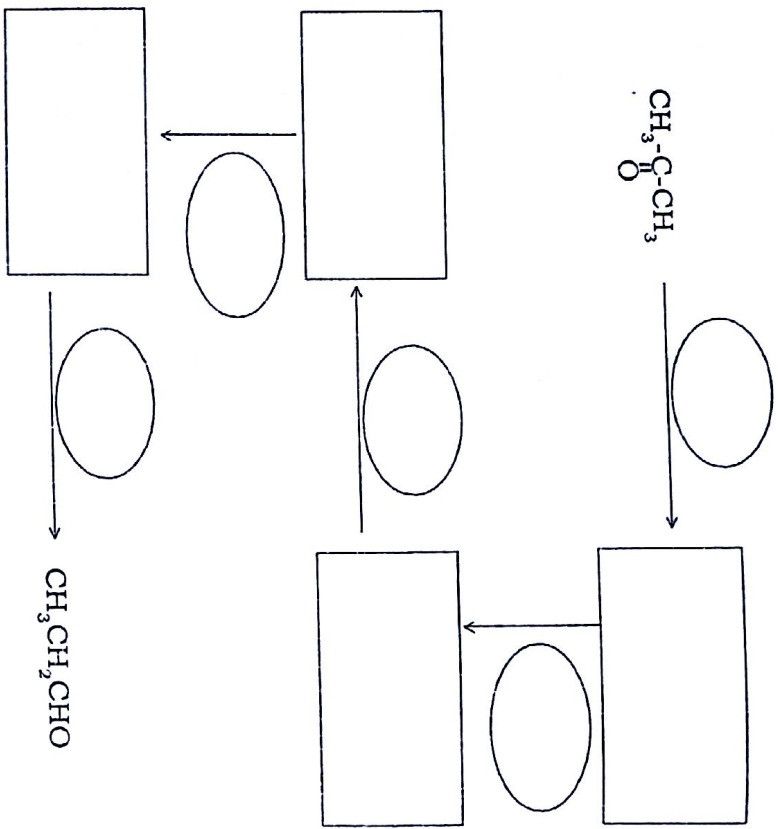


13. List of the reagent and solvent:

PCl₅, NH₃,
 K₂Cr₂O₇, dil H₂SO₄,
 C₂H₅OH, water, dry (C₂H₅OC₂H₅)



14. List of the reagent and solvent:
 Conc. H_2SO_4 , HBr , $NaBH_4$,
 aq. $NaOH$, dry $(C_2H_5OC_2H_5)$, Peroxybenzoic acid,
 Pyridiniumchlorochromate(PCCl),

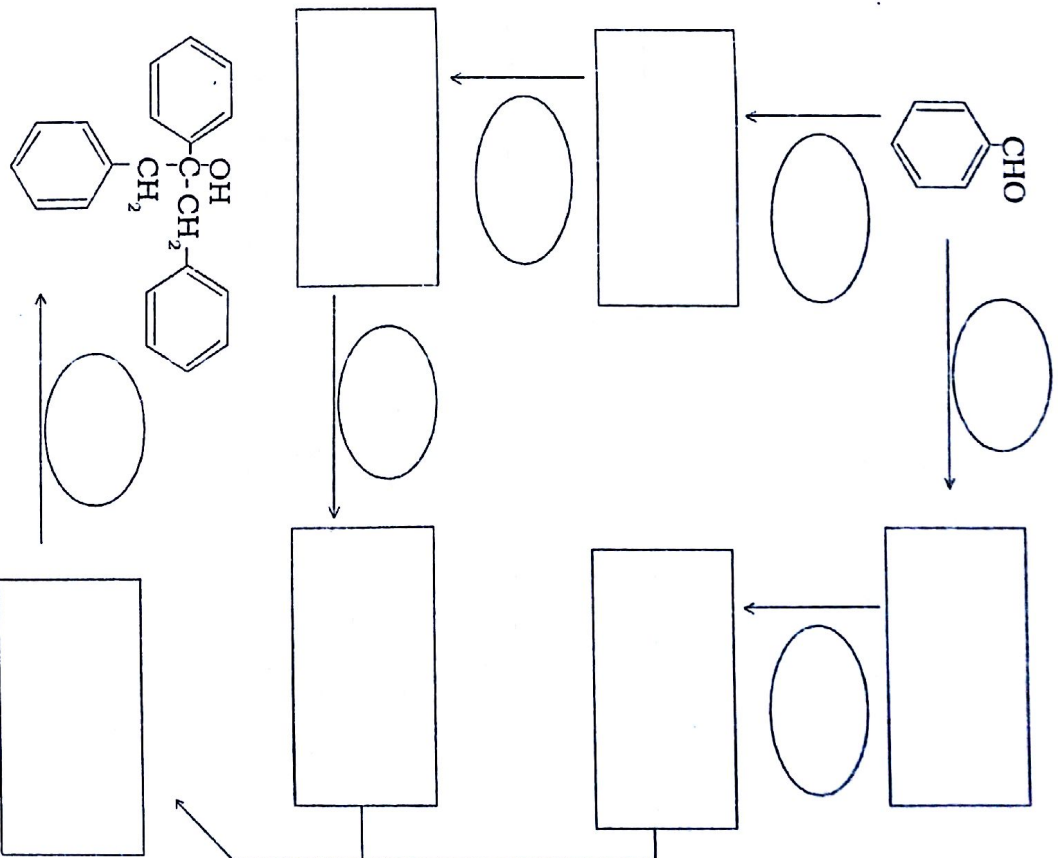


List of the reagent and solvent:

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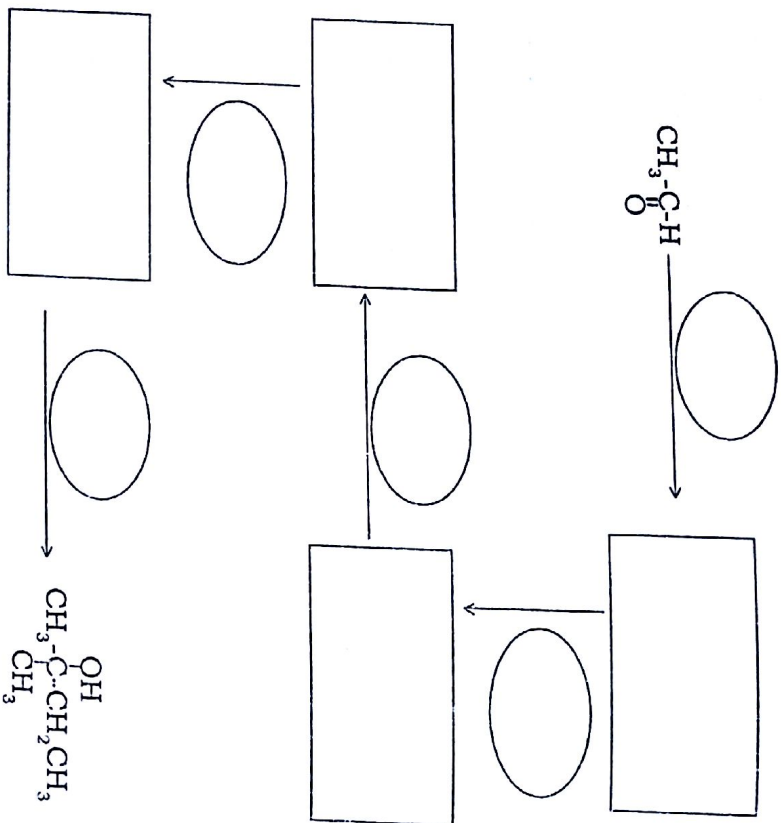
15. List of the reagent and solvent:
 Mg , PCl_5 , $K_2Cr_2O_7$, dil. H_2SO_4 , $LiAlH_4$,
 C_2H_5OH , water, dry $(C_2H_5OC_2H_5)$



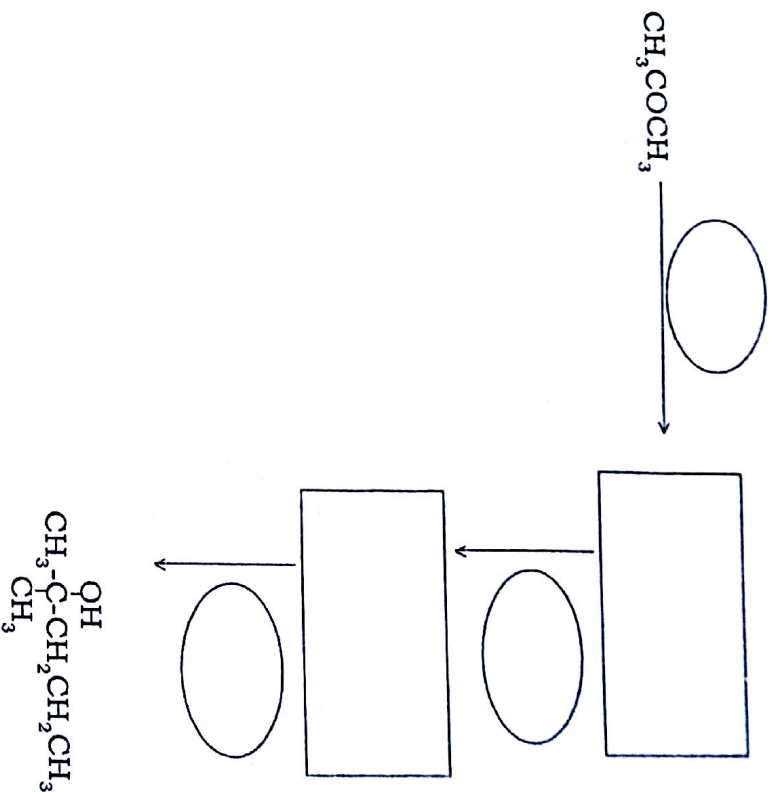
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16. List of the reagent and solvent:
 Mg, PCl₅, dil HCl, LiAlH₄,
 CH₃COCH₃, C₂H₅OH, water, dry (C₂H₅OC₂H₅)

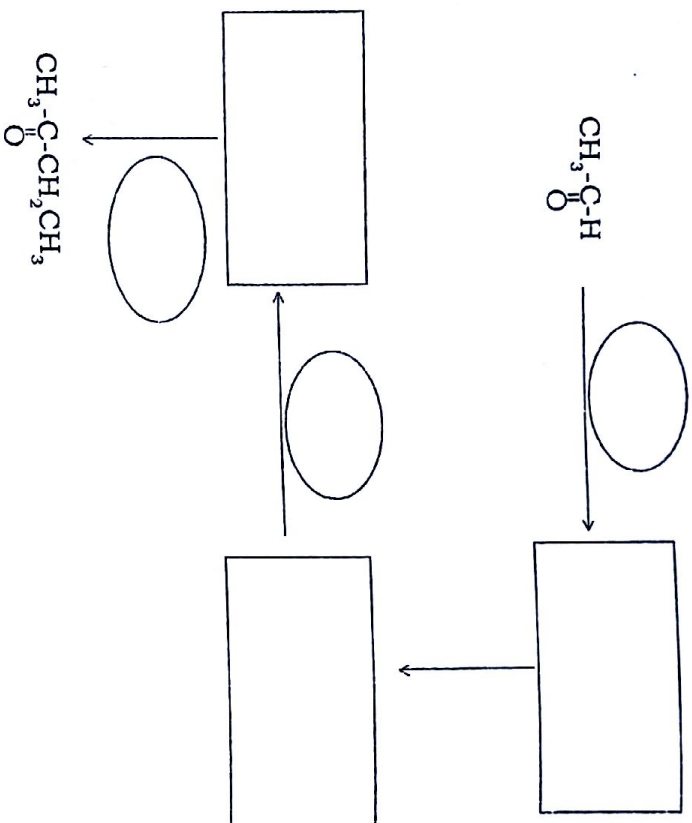


17. List of the reagent and solvent:
 Zn, Hg, Conc.HCl, LiAlH₄, aq. NaOH, Conc.H₂SO₄,



18. List of the reagent and solvent:

Mg, Zn, Hg
HBr, $K_2Cr_2O_7$, dil H_2SO_4 , Conc. HCl
aq NaOH, C_2H_5OH , water, dry ($C_2H_5OC_2H_5$)

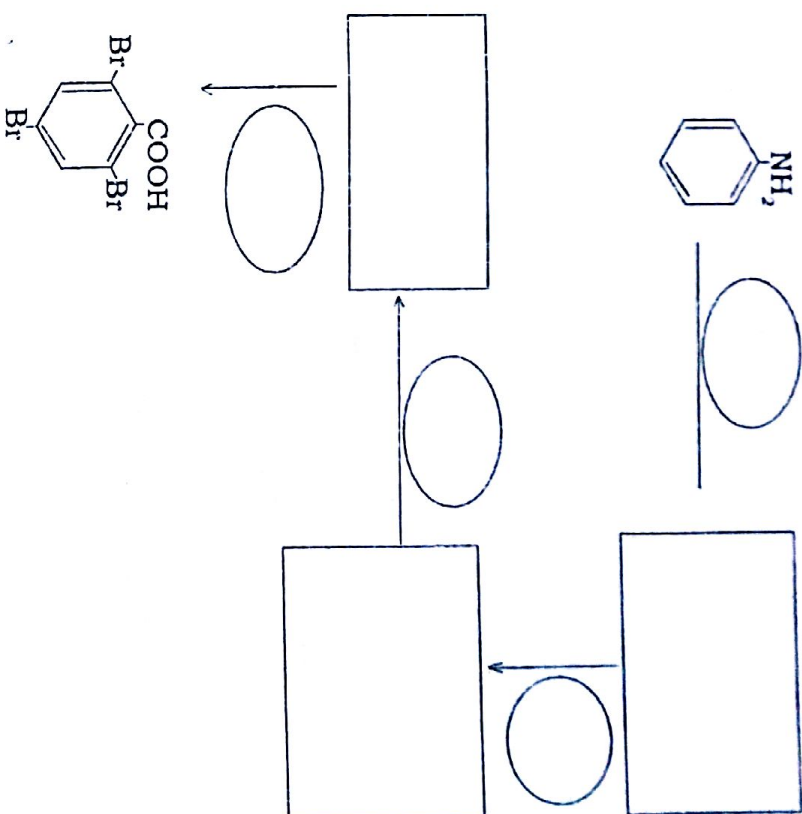


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19. List of the reagent and solvent:

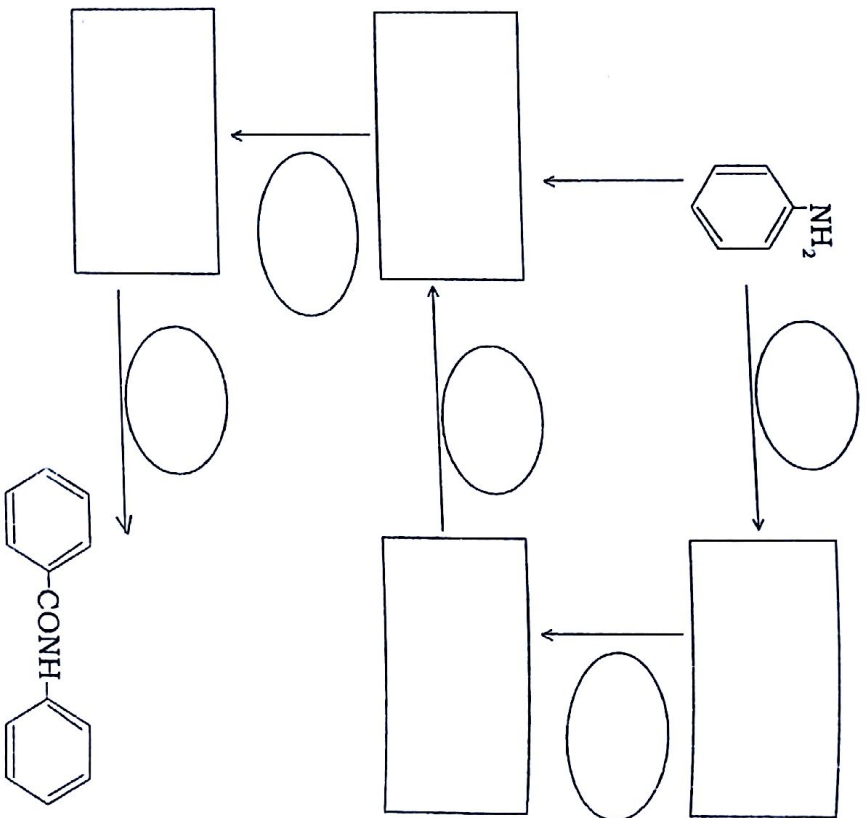
$NaNO_2$, LiAlH₄, KCN
Br₂, dil. HCl, CuCN
water, dry ($C_2H_5OC_2H_5$)



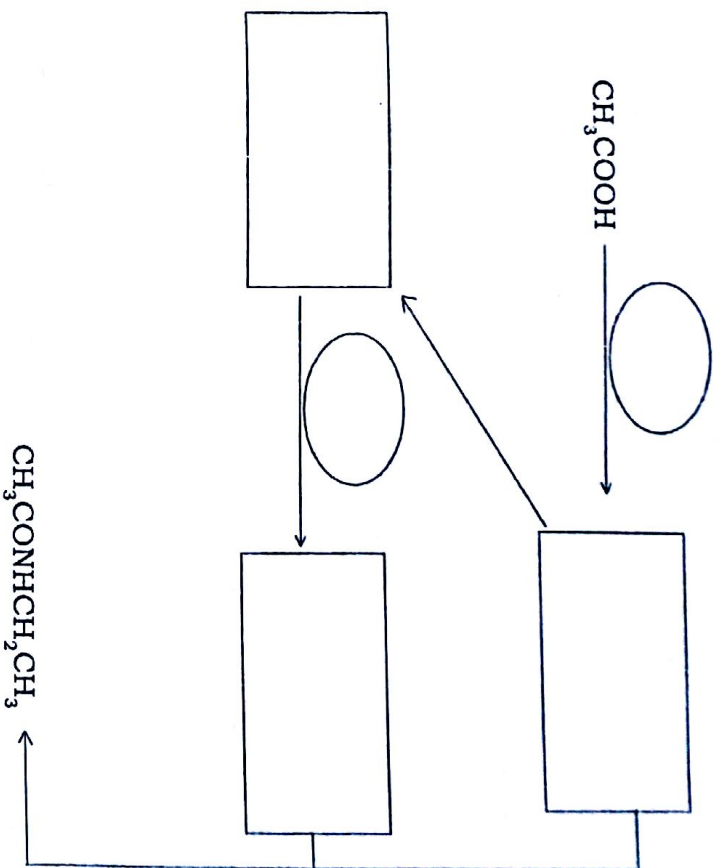
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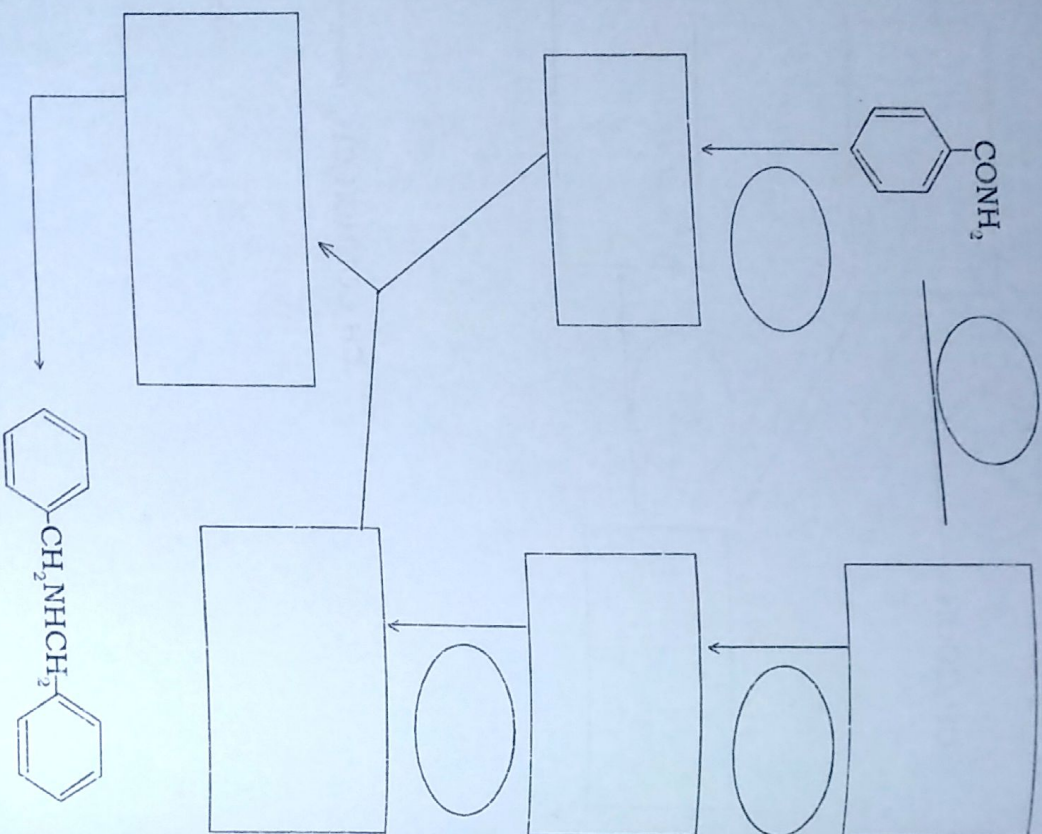
20. List of the reagent and solvent:
 NaNO_2 , dil. HCl , CuCN , KCN , PCl_5
 $\text{C}_2\text{H}_5\text{OH}$, water, dry ($\text{C}_2\text{H}_5\text{OC}_2\text{H}_5$)



21. List of the reagent and solvent:
 PCl_5 , dil HCl , LiAlH_4 , NH_3
 water, dry ($\text{C}_2\text{H}_5\text{OC}_2\text{H}_5$)



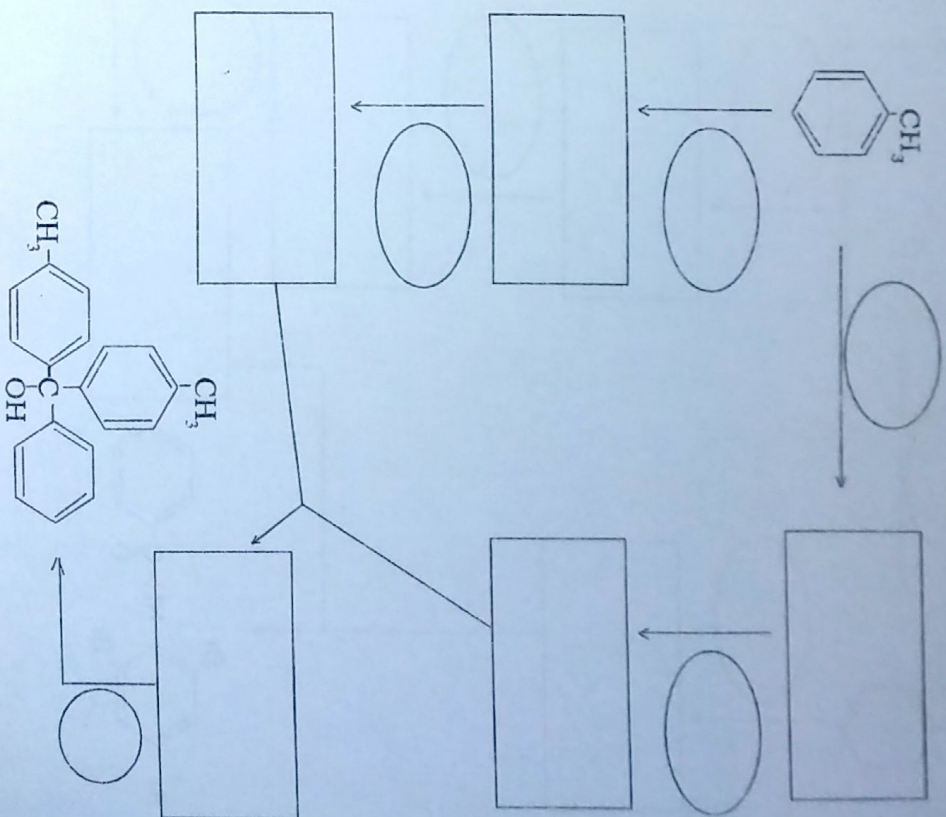
22. List of the reagent and solvent:
 PCl_5 , LiAlH_4 , NaOH , dil. HCl
 $\text{C}_2\text{H}_5\text{OH}$, water, dry ($\text{C}_2\text{H}_5\text{OC}_2\text{H}_5$)



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23. List of the reagent and solvent:
 Mg , Zn , Pt , Fe , Hg
 Br_2 , PCl_5 , AlCl_3
 KCN , $\text{K}_2\text{Cr}_2\text{O}_7$, Conc. HNO_3 , $\text{Conc. H}_2\text{SO}_4$, Conc. HCl
 aq. NaOH ,
 CH_3COCH_3 , $\text{C}_2\text{H}_5\text{OH}$, water, dry ($\text{C}_2\text{H}_5\text{OC}_2\text{H}_5$)



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24. List of the reagent and solvent:

Br_2 , PCl_5 , AlCl_3

NaNO_2 , KCN , dil.HCl , aq NaOH , CuCN

$\text{C}_2\text{H}_5\text{OH}$, water, dry ($\text{C}_2\text{H}_5\text{OC}_2\text{H}_5$)

